A NOVEL SYNTHESIS OF THE (2R,3S)- AND (2S,3R)-3-AMINO-2-HYDROXYCARBOXYLIC ACID DERIVATIVES, THE KEY COMPONENTS OF A RENIN INHIBITOR AND BESTATIN, FROM METHYL (R)- AND (S)-MANDELATE

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Abstract: The title synthesis could be accomplished by featuring the [2+2]-cycloaddition reaction of a chiral imine with benzyloxyketene, alcoholysis of the formed 2-azetidinone derivative, and reductive removal of the mandelate-derived benzylic oxygen by way of a 2-oxazolidone derivative.

Optically active 3-amino-2-hydroxycarboxylic acid derivatives are often involved in medicinally important compounds as their key components. Thus, one of the promising renin inhibitor $(1)^{2}$) bears isopropyl (2R,3S)-3-amino-4-cyclohexyl-2-hydroxybutyrate $(2)^{2,3}$ as its C-terminal moiety, and bestatin(3), the famous immunological response modifier,⁴) consists of (2S,3R)-3-amino-2-hydroxy-4-phenylbutyric acid $(4)^{3d,5}$ and (S)-leucine.

We wish to report here a novel synthesis of these antipodal compounds (2 and 4) starting from methyl (R)-and (S)-mandelate (5 and ent-5). The explored synthetic scheme features the [2+2]-cycloaddition reaction of a chiral imine with benzyloxyketene, alcoholysis of the formed 2-azetidinone derivative, and reductive removal of the mandelate-derived benzylic oxygen by way of a 2-oxazolidone derivative. It was previously disclosed that the [2+2]-cycloaddition reaction of chiral imine having an asymmetric center on the nitrogen atom of C=N bond with benzyloxyketene was unrewarding for the preparation of 2 because of its low diastereoselectivity.^{3c)} However, we have now found that a chiral imine derived from 5 or ent-5, which bears an asymmetric center on the carbon atom of C=N bond, can react with benzyloxyketene in a highly diastereoselective manner and the 2-azetidinone derivative produced as a major addition product can be ingeniously elaborated to 2 or 4.



As shown in **Scheme 1**, the synthesis of 2 commences with protection of the hydroxy group of 5 with *t*-butyldimethylsilyl (TBDMS) or *t*-butyl (*t*-Bu) group.⁶) These protective groups were employed since they can be readily removed under the conditions for acidic alcoholysis of a 2-azetidinone derivative (*vide infra*). Reduction of the protected esters (6 and 7) with diisobutylaluminum hydride (DIBAL) smoothly produced the corresponding aldehydes (8 and 9).⁶) Two sorts of the aldehydes (8



a) TBDMSCl, ImH, DMF, rt, overnight, 98% (6) or Me₂C=CH₂, conc.-H₂SO₄, CH₂Cl₂, rt, 2 days, 83% (7) b) DIBAL, Et₂O-C₆H₁₄, -78°C, 20 min, 82% (8) or 30 min, 73% (9) c) DAMNH₂ or BnNH₂, anhyd. MgSO₄, PhMe, 0°C, 50 min (10a) or 1 h (10b-d) (see Table 1) d) BnOCH₂COCl (1.3 equiv.), Et₃N (1.5 equiv.), CH₂Cl₂, rt, overnight (see Table 1) e) HCl, *i*-PrOH, rt, overnight, then 60°C, 3 h (11a→13), 50°C, 3 h (11b→14), 40°C, 7 h (11c→13), or 40°C, 5 h (11d→13) (see Table 1) f) Cl₃COCOCCl, pyridine, CH₂Cl₂, 0°C, 10 min (13→15), or 40°C, 4.5 h (14→16) (see Table 1) g) H₂, 10% Pd/C, EtOAc, rt, overnight (see Table 1) h) H₂ (5 atom), 5% Rh/Al₂O₃, AcOH, 97%

and 9) were condensed with di-p-anisylmethylamine (DAMNH₂) or benzylamine (BnNH₂) in the presence of anhydrous magnesium sulfate, giving rise to four types of the chiral imines (10a-d). The [2+2]-cycloaddition reactions of 10a-d with benzyloxylketene *in situ* produced from benzyloxyacetyl chloride in the presence of triethylamine, proceeded in a highly diastereoselective manner, yielding mixtures of the 3,4-*cis*-2-azetidinone derivatives (11a-d and 12a-d) in which the desired diastereomers (11a-d) were highly predominent.^{6,7}) As summarized in Table 1, the best chemical yield (88%) and diastereoselectivity (15:1) could be realized for the reactions employing 10a and 10d as chiral imines, respectively. Considering the chemical yields of the later synthetic stages as well as the chemical yield and diastereoselectivity of the [2+2]-cycloaddition reaction, the reaction of 10a with benzyloxyketene seems to be most practical. Stereochemistries of 11a-d and 12a-d could be definitely assigned by their ¹H-NMR spectra⁹) in addition to successful syntheses of 2 from 11a-d (*vide infra*)¹¹)

Elaboration of the major products (11a-d) to 2 could be achieved in 4 steps. Thus, treatments of 11a,c with acidic isopropanol (*i*-PrOH) cleanly effected simultaneous alcoholysis of the 2-azetidinone moieties and removals of the TBDMS or the *t*-Bu group and the DAM groups, affording the same isopropyl ester (13).⁶ It is noteworthy that the DAM group could be readily cleft under the simple acidic conditions. When 11b,d were subjected to the same conditions as employed for 11a,c, the isopropyl ester (14) could be produced without cleavage of the N-benzyl group. In order to

	_		Yield (%)				
	R ²	R ³	8 or 9→10	$10 \rightarrow 11$ and 12	11-→13 or 14 ^d)	13→15 or	15 or 16→17
				(11:12)		14-→16 ^{d)}	
a	TBDMS	DAM	100	88 (10:1) ^{b)}	84 (13) ^{e)}	90 (15)	94
b	TBDMS	Bn	100	59 (12:1) ^{b)}	86 (14) ^{e)}	64 (16)	81
с	t-Bu	DAM	100	77 (9:1) ^{c)}	59 (13) ^{f)}		<u> </u>
d	t-Bu	Bn	98	62 (15:1) ^{c)}	69 (14) ^f)		` `

Table 1 Chemical Yields of Imine Formation (8 or $9\rightarrow10$), [2+2]-Cycloaddition (10 $\rightarrow11$ and 12), Alcoholysis (11 $\rightarrow13$ or 14), 2-Oxazolidone Formation (13 $\rightarrow15$ or 14 $\rightarrow16$), and Hydrogenolysis (15 or 16 $\rightarrow17$)^{a)}

a) The reaction conditions are given in the footnotes of Scheme 1. b) Determined by weighing separated 11 and 12. c) Determined by the ¹H-NMR spectrum of the mixture of 11 and 12. Separation of 11 and 12 could not be achieved. d) Numbers in parentheses indicate the compound numbers. e) A pure sample of 11 was used for the reaction. f) A mixture of 11 and 12 was directly used for the reaction. The yield was calculated based on the total amount of the mixture.

remove the benzylic hydroxyl groups derived from 5, two sorts of the isopropyl esters (13 and 14) were converted to the 2-oxazolidone derivatives (15 and 16)⁶) by treating with trichloromethyl chloroformate (phosgen dimer) in the presence of pyridine. Hydrogenolyses of 15 and 16 over 10% Pd on charcoal afforded same isopropyl (2*R*,3*S*)-3-amino-2-hydroxy-4-phenylbutyrate (17).⁶) The convergent syntheses of 17 could nicely correlate the stereochemistries of 11a-d. Further catalytic reduction of 17 over 5% Rh on alumina furnished optically pure 2, mp 86-86.5°C and $[\alpha]_D^{20}$ -22.0° (c=1.18, CHCl₃) [*lit.*,^{3c}) mp 86-87°C and $[\alpha]_D^{20}$ -22.0° (c=1.08, CHCl₃)].

With completion of the synthesis of 2 from 5, the explored synthetic route was next applied to ent-5 to prepare 4. Thus, as shown in Scheme 2, a mixture of ent-11a and ent-12a (8:1) could be prepared similarly from ent-5 in 4 steps. A combination of TBDMS and DAM groups was employed to obtain a higher yield in the [2+2]-cycloaddition reaction. Elaboration of ent-11a to ent-17 according to the same procedure as discribed above gave ent-17. Acidic hydrolysis of ent-17 followed by treatment with an ion exchange resin, afforded 4, mp 235-237°C (decomp.) and $[\alpha]_D^{20}$ +29.9° (c=0.214, 1M HCl) [*lit.*,⁴) mp 219-221°C and $[\alpha]_D^{20}$ +27.9° (c=0.717, 1M HCl)].¹²)

As mentioned above, we have succeeded in exploring a novel synthetic route to antipodal 2 and 4. The best combined yields (overall 8 steps) of 2 and 4 from 5 and ent-5 by way of 11a and ent-11a can be calculated as 44% and 30%, respectively. Taking into account high diastereoselectivity observed for the [2+2]-cycloaddition reaction, expeditious elaboration of 11 or ent-11 to 2 or 4, and use of com-

Scheme 2



a) For reaction conditions [steps i)-vil)], see footnotes of Scheme 1. i) 93% ii) 80% iii) 100% iv) 90% (ent-11a:ent-12a = 8:1) v) 70% vi) 93% vii) 92% viii) 6M HCl, 100°C, 4h, then ion exchange resin (AG 50 X W2, H⁺-form), 85%

mercially available 5 or ent-5 as a starting material, the overall process may have potential as one of the most reliable synthetic methods for preparing the (2R,3S)- and (2S,3R)-3-amino-2hydroxycarboxylic acid derivatives such as 2 and 4.

References and Notes

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- 6) Representative physical data of the synthetic intermediates are as follows. 6: mp 40-41°C, $[\alpha]_D^{20}$ -50.0° (c=1.04, CHCl₃); 8: oil; 10a: caramel; 11a: caramel, [α]D²⁰ +32.4° (c=1.11, CHCl₃); 12a: mp 96-97°C, [a]D²⁰ +2.5° (c=1.22, CHCl₃); 13: oil; 15: oil, [a]D²⁰ +99.7° (c=1.62, CHCl₃); 17: mp 112.5-113°C, $[\alpha]_D^{20}$ -32.7° (c=1.04, CHCl₃).
- 7) The precise reaction mechanism which may rationalize the preferential formations of 11a-d is quite ambiguous. However, as previously suggested for the similar [2+2]-cycloaddition reaction,⁸⁾ the observed results may be accounted for by initial formation of the zwitter-ionic intermediates (18) from 10a-d and benzyloxyketene and subsequent conrotatory ring closure to the indicated direction under an influence of the adjacent chiral center.
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- 9) The coupling constants of the C3- and C4-protons in the 2-azetidinone rings of 11a-d and 12a-d were found to be 4.9-5.3 Hz. Since $C_{3,4}$ -trans-2-azetidinone derivatives regularly exhibit $J_{3,4} = -2.5$ Hz, ¹⁰⁾ these spectral characteristics clearly suggest that both 11a-d and 12a-d bears the C_{3,4}-cis stereochemistries.

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stries was further ascertained by the following chemical correlation. Thus, desilylation of 11a [Bu4NF, THF, 99%] followed by Swern oxidation of the alcohol (19) [(COCl)₂-DMSO-Et₃N,CH₂Cl₂, 100%] and reduction of the ketone (20) [NaBH₄, MeOH-CH₂Cl₂, 87%] gave a mixture of the diastereomeric alcohols (19 and 21, **19:21**=1:10). The predominantly produced alcohol (**21**),



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 $[\alpha]_D^{20}$ +26.1° (c=1.31, CHCl₃), was found to be enantiomeric to the alcohol, $[\alpha]_D^{20}$ -26.9° (c=1.20, CHCl3), independently prepared by desilylation of 12a [Bu4NF, THF, 71%].

12) The hydrochloride of 4 showed mp 190-192°C (decomp.) [lit., ^{3d}) mp 190°C (decomp.)].

(Received in Japan 5 March 1990)